# Oxocarbons and related compounds. Part 25.<sup>1</sup> Heterocycle-fused benzocyclobutenediones. Synthesis of indolo-, benzofuro- and benzothieno-dihydrobenzocyclobutenediones and benzocyclobutenediones

# PERKIN

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A general and efficient procedure has been described for the synthesis of heterocycle-fused dihydrobenzocyclobutenediones and benzocyclobutenediones by reacting (alk-1-enyl)heterocycles with semisquaric chloride.

Substituted cyclobutenediones <sup>2</sup> and benzocyclobutenediones <sup>2</sup> have recently attracted much attention as synthons as well as in the construction of complex organic molecules.<sup>3,4</sup> By contrast, only a small number of carbocycle-fused benzocyclobutenediones <sup>5</sup> (CBBDS) <sup>5j,6</sup> have been presented in the literature and, to the best of our knowledge, only one heterocycle-fused benzocyclobutenedione (HBBD) has been reported.<sup>7</sup>

Recently <sup>5j</sup> we have reported that (alk-1-enyl)arenes readily undergo Diels-Alder reactions with 3-chlorocyclobut-3-ene-1,2-dione (semisquaric chloride)  $1^8$  to give dihydrocyclobuta-[*a*]naphthalene-1,2-diones, which were subsequently dehydrogenated to cyclobuta[*a*]naphthalene-1,2-diones. For potentially wider applicability, we have now extended this procedure, developing a simple, straightforward synthesis of the heterocycle-fused dihydrobenzocyclobutenediones C (Scheme 1). This involves the Diels-Alder reaction of an (alk-1-enyl)hetero-



cyclic compound A, containing a conjugated diene system, with semisquaric chloride 1, followed by regiospecific elimination of hydrogen chloride to give B and subsequent allylic rearrangement to furnish C. Dehydrogenation of C should then afford the desired heterocycle-fused benzocyclobutenedione D.

When the 2-(alk-1-enyl)-N-methylindoles 2a and 2b were heated with semisquaric chloride 1 in dichloromethane, the corresponding dihydrocyclobuta[c]carbazole-1,2-diones 3aand 3b were isolated from the reaction mixture by column chromatography (Scheme 2). In addition a chlorine-containing



Scheme 2 Synthesis of heterocycle-fused dihydrobenzocyclobutenediones and benzocyclobutenediones. For conditions see Table 1.

side-product was obtained in each case. The structures **5a** and **5b** were attributed to these. Similarly, the 2-(alk-1-enyl)-1benzofurans **2c**, **2d** and 2-(prop-1-enyl)-1-benzothiophene **2e** reacted with semisquaric chloride **1** to give the benzofuro- and benzothieno-fused dihydrobenzocyclobutenediones **3c**, **3d** and **3e**, respectively (Scheme 2, Table 1).

In addition, 3-(prop-1-enyl)-1-benzothiophene 6 was also treated with semisquaric chloride 1 to give 7, an isomer of 3e (Scheme 3). The heterocycle-fused dihydrobenzocyclo-



**Scheme3** Synthesis of 3-methyl-3,4-dihydrobenzo[b]cyclobuta[e][1]-benzothiophene-1,2-dione and its dehydrogenation. For conditions see Table 1.

butenediones 3c, 3d, 3e and 7 were easily dehydrogenated by treatment with bromine (*ca.* 1 equiv.) in boiling tetrachloromethane. The benzofuro- and benzothieno-fused benzocyclobutenediones 4c, 4d, 4e and 8 thus obtained, are the first representatives of this class of compounds (Schemes 2 and 3).

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 Table 1
 Indolo-, benzofuro- and benzothieno-fused dihydrobenzocyclobutenediones

Substrate	Solvent	Conditions (temp., <i>t</i> /h)	Product <sup>a</sup>	Yield <sup> b</sup> (%)	Мр <sup><i>c</i></sup> ( <i>T</i> /°С)
2a	CH <sub>2</sub> Cl <sub>2</sub>	rt, 20 h; then reflux, 1 h	3a	67	249–251
2b	CH <sub>2</sub> Cl <sub>2</sub>	rt, 20 h; then reflux, 1 h	3b	30	177–178
2c	CH <sub>2</sub> Cl <sub>2</sub>	rt, 48 h; then 60 °C, 0.5 h	3c	38	188–190 (decomp.)
2d	Neat	rt, 1 h; then 40 °C, 1 h; then 60 °C, 1.5 h	3d	45	176–178 (decomp.)
2e	CH <sub>2</sub> Cl <sub>2</sub>	rt, 20 h; then reflux, 1 h	3e	35	167–168 (decomp.)
6	CH <sub>2</sub> Cl <sub>2</sub>	rt, 24 h; then reflux, 1 h	7	56	205–207 (decomp.)
3c	CCl <sub>4</sub>	reflux, 5 h	4c	76	218–220 (decomp.)
3d	CCl <sub>4</sub>	reflux, 5 h	4d	59	(decomp.) (decomp.)
3e 7	CCl <sub>4</sub> CCl <sub>4</sub>	reflux, 4 h reflux, 5.5 h	4e 8	51 88	219–220 253–255

<sup>a</sup> All compounds gave satisfactory elemental analyses and were characterized by IR, <sup>1</sup>H NMR, <sup>13</sup>C NMR and mass spectroscopy. <sup>b</sup> Yields are of purified products. <sup>c</sup> Uncorrected.

### Experimental

### 3-Methyl-3,4-dihydrobenzo[b]cyclobuta[e][1]benzofuran-1,2dione 3d

Typical procedure. Semisquaric chloride 1 (1.16 g, 10 mmol) and 2-(prop-1-enyl)-1-benzofuran 2d (1.58 g, 10 mmol; corrected for (E)-2d: 1.30 g, 8.23 mmol) were combined, and the mixture was stirred at rt for 60 min during which time the reaction mixture took on a dark-red colour. It was then heated at 40 °C for 1 h, and at 60 °C for 1.5 h. Liberated HCl was removed under reduced pressure. The resulting dark brown crystals were triturated with ethanol. The yellow crystals obtained were recrystallized from ethanol to give 3d (0.88 g, 45%), mp 176–178 °C (decomp.) (Found: C, 75.6; H, 4.3.  $C_{15}H_{10}O_3$  requires C, 75.6; H, 4.2);  $\nu_{max}(KBr)/cm^{-1}$  2960, 1780–1760, 1600, 1545, 1160 and 765;  $\delta_{\rm H}$ (400 MHz; CDCl<sub>3</sub>) 1.50 (3 H, d, J 7.3 ‡), 2.95 (1 H, dd, J 8.6 and 18.6), 3.40 (1 H, dd, J 9.9 and 18.5), 3.62 (1 H, m), 7.30 (2 H, m), 7.48 (1 H, m) and 7.76 (1 H, m);  $\delta_{\rm C}(100.6 \text{ MHz}; \text{CDCl}_3)$  18.1, 29.7, 30.7, 109.5, 111.8, 120.7, 122.4, 125.6, 155.9, 162.2, 189.1, 192.8, 193.1 and 196.1; m/z (EI) 238 (M<sup>+</sup>, 55%), 210 (M<sup>+</sup> - CO, 49), 182  $(M^+ - 2CO, 79)$  and 181 (100).

## 3-Methylbenzo[b]cyclobuta[e][1]benzofuran-1,2-dione 4d

To a suspension of compound 3d (0.48 g, 2.01 mmol) in CCl<sub>4</sub> (50 cm<sup>3</sup>) was added a solution of bromine (0.31 g, 1.91 mmol) in CCl<sub>4</sub> (20 cm<sup>3</sup>). The mixture was stirred and heated at reflux until no further HBr was generated (*ca.* 5 h). On cooling to -15 °C compound 4d precipitated and it was filtered off and

recrystallized from ethanol to give yellow crystals (0.28 g, 59%), mp 198–200 °C (decomp.) (Found: C, 76.0; H, 3.6.  $C_{15}H_8O_3$ requires C, 76.3; H, 3.4);  $v_{max}(KBr)/cm^{-1}$  1780–1755, 1600, 1450, 1265, 1185, 1110 and 750;  $\delta_H(400 \text{ MHz; CDCl}_3)$  2.78 (3 H, s), 7.45 (1 H, t, J 7.6), 7.60 (3 H, m) and 8.17 (1 H, d, J 7.7);  $\delta_C(100.6 \text{ MHz; CDCl}_3)$  18.3, 112.3, 116.9, 119.7, 121.6, 123.3, 124.7, 129.5, 134.1, 157.4, 159.4, 167.3, 169.0, 193.2 and 193.7; m/z (EI) 236 (M<sup>+</sup>, 7%), 208 (M<sup>+</sup> – CO, 19) and 180 (M<sup>+</sup> – 2CO, 100).

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### References

- 1 Part 24, A. H. Schmidt, S. H. Thiel and O. Gaschler, J. Chem. Soc., Perkin Trans. 1, preceding paper.
- 2 For reviews, see: (a) G. Maahs and P. Hegenberg, Angew. Chem., 1966, **78**, 927; Angew. Chem., Int. Ed. Engl., 1966, **5**, 888; (b) H. E. Sprenger and W. Ziegenbein, Angew. Chem., 1968, **80**, 541; Angew. Chem., Int. Ed. Engl., 1968, **7**, 530; (c) W. Ried and A. H. Schmidt, Angew. Chem., 1972, **84**, 1048; Angew. Chem., Int. Ed. Engl., 1972, **11**, 997; (d) A. H. Schmidt and W. Ried, Synthesis, 1978, 1; (e) H. Knorr and W. Ried, Synthesis, 1978, 649; (f) A. H. Schmidt and W. Ried, Synthesis, 1978, 869; (g) A. H. Schmidt, Synthesis, 1980, 961.
- 3 (a) M. E. Jung and J. A. Lowe, J. Org. Chem., 1977, 42, 2371; (b) L. A. Spangler and J. S. Swenton, J. Org. Chem., 1984, 49, 1800 and references cited therein; (c) C. F. Wilcox, Jr. and E. N. Farley, J. Org. Chem., 1985, 50, 351; (d) L. S. Liebeskind, S. L. Baysdon, M. S. South, S. Iyer and J. P. Leeds, Tetrahedron, 1985, 41, 5839 and references cited therein; (e) L. S. Liebeskind, S. Iyer and C. F. Jewell, Jr., J. Org. Chem., 1986, 51, 3065; (f) S. T. Perry, L. D. Foland, O. H. W. Decker and H. W. Moore, J. Org. Chem., 1986, 51, 3067; (g) O. H. W. Decker and H. W. Moore, J. Org. Chem., 1987, 52, 1174; (h) L. D. Foland and H. W. Moore, J. Am. Chem. Soc., 1989, 111, 989; (i) L. S. Liebeskind, J. Am. Chem. Soc., 1990, 112, 291.
- 4 For a review on the construction of complex organic molecules by ring expansion of 4-hydroxycyclobut-2-en-1-ones, which are easily obtainable from cyclobutenediones, see: H. W. Moore and B. R. Yerxa, *Chemtracts Org. Chem.*, 1992, **5**, 273.
- 5 (a) M. P. Cava and B. Hwang, Tetrahedron Lett., 1965, 2297; (b)
  J. F. W. McOmie and D. H. Perry, J. Chem. Soc., Chem. Commun., 1973, 248; (c) A. C. Hsu and M. P. Cava, J. Org. Chem., 1979, 44, 3790; (d) K. J. Gould, N. P. Hacker, J. F. W. McOmie and D. H. Perry, J. Chem. Soc., Perkin Trans. 1, 1980, 1834; (e) N. P. Hacker, J. F. W. McOmie, J. Meunier-Piret and M. Van Meersche, J. Chem. Soc., Perkin Trans. 1, 1983, 2659; (f) R. F. C. Brown, K. J. Coulston, F. W. Eastwood and S. Saminathan, Aust. J. Chem., 1987, 40, 107; (g)
  R. F. C. Brown, N. R. Browne, K. J. Coulston and F. W. Eastwood, Aust. J. Chem., 1990, 43, 1935; (h) M. Adeny, R. F. C. Brown, K. J. Coulston, F. W. Eastwood and I. W. James, Aust. J. Chem., 1991, 44, 967; (i) A. H. Schmidt, Ch. Künz, M. Malmbak and J. Zylla, Synthesis, 1994, 422; (j) A. H. Schmidt, G. Kircher, Ch. Künz, S. Wahl and M. W. Hendriok, J. Org. Chem., 1995, 60, 3890.
- 6 P. R. Buckland, N. P. Hacker and J. F. W. McOmie, J. Chem. Soc., Perkin Trans. 1, 1983, 1443.
- 7 L. S. Liebeskind, L. J. Lescosky and C. M. McSwain, J. Org. Chem., 1989, 54, 1435.
- 8 A. H. Schmidt, M. Debo and B. Wehner, Synthesis, 1990, 237.

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<sup>‡</sup> J Values given in Hz.